

1۔ ہر سوال کے سامنے چار دائرے دئے گئے ہیں، صرف صحیح جواب والا دائرہ بھریں۔

2۔ دائروں کو شیڈ (بھرنے) کے لئے نیلے یا کالے رنگ کا مارکر استعمال کریں۔

3۔ جواب میں ایک سے زائد دائرے بھرنے سے جواب غلط تصور ہوگا۔

Time Allowed: 20 Minutes

SECTION – A

Marks : 18

1	The largest no of molecules is present in	<input type="radio"/> 22 g of CO ₂	<input type="radio"/> 49 g of H ₂ SO ₄	<input type="radio"/> 18 g of H ₂ O	<input type="radio"/> 90 g of C ₆ H ₁₂ O ₆
2	The mass of one molecule of H ₂ O is	<input type="radio"/> 18 g	<input type="radio"/> 0.18 g	<input type="radio"/> $\frac{18}{6.02 \times 10^{23}}$ g	<input type="radio"/> $\frac{6.02 \times 10^{23}}{18}$ g
3	How many pairs of electrons surround the central atom of "Sn" in the angular SnCl ₂ molecule?	<input type="radio"/> 1	<input type="radio"/> 2	<input type="radio"/> 3	<input type="radio"/> 4
4	The energy carried by a photon is given by:	<input type="radio"/> $E = h\nu$	<input type="radio"/> $E = \frac{hc}{\lambda}$	<input type="radio"/> Both A and B	<input type="radio"/> None of these
5	The e/m ration for canal rays particles is than that of cathode ray particles.	<input type="radio"/> Smaller	<input type="radio"/> Greater	<input type="radio"/> Equal	<input type="radio"/> May be greater or smaller
6	Which bonds are responsible for hair to be straight or curly?	<input type="radio"/> Hydrogen bonding	<input type="radio"/> Disulphide bridge	<input type="radio"/> Glycosidic linkage	<input type="radio"/> Peptide linkage
7	Which of the given gases can be liquefied above its critical temperature?	<input type="radio"/> O ₂	<input type="radio"/> N ₂	<input type="radio"/> H ₂	<input type="radio"/> None of these
8	If the mass of electron is doubled, the Rydberg constant:	<input type="radio"/> Becomes half	<input type="radio"/> Becomes double	<input type="radio"/> Remains unchanged	<input type="radio"/> Becomes one fourth
9	Hydrolysis of a salt may result in solution.	<input type="radio"/> Acidic	<input type="radio"/> Basic	<input type="radio"/> Neutral	<input type="radio"/> Any of the given
10	The value of p _{kw} with increase in T.	<input type="radio"/> Decreases	<input type="radio"/> Increases	<input type="radio"/> Remains constant	<input type="radio"/> Both A & B
11	A crystal can have centre / centres of symmetry at maximum.	<input type="radio"/> 1	<input type="radio"/> 2	<input type="radio"/> 3	<input type="radio"/> 4
12	The rate of evaporation of a liquid increases by	<input type="radio"/> Increasing surface area	<input type="radio"/> Decreasing intermolecular forces	<input type="radio"/> Increasing temperature	<input type="radio"/> All these
13	Rate = K[A][B]. This mathematical expression is called	<input type="radio"/> Rate law	<input type="radio"/> Rate expression	<input type="radio"/> Rate equation	<input type="radio"/> All these
14	Daniell cell is not cell.	<input type="radio"/> An electrochemical	<input type="radio"/> A Galvanic	<input type="radio"/> An electrolytic	<input type="radio"/> A voltaic
15	Which of the given is not a colligative property of a solution?	<input type="radio"/> Lowering of vapour pressure	<input type="radio"/> Elevation in B.P°	<input type="radio"/> Osmotic pressure	<input type="radio"/> Mole fraction
16	Which of the given is the most strongest?	<input type="radio"/> Hydrogen bonding	<input type="radio"/> Dipole-dipole interaction	<input type="radio"/> London dispersion forces	<input type="radio"/> Intramolecular forces
17	Pepsin is found in	<input type="radio"/> Salvia	<input type="radio"/> Gastric Juice	<input type="radio"/> Bile extract	<input type="radio"/> Pancreatic juice
18	For an ideal gas the compressibility factor has a value of	<input type="radio"/> Zero	<input type="radio"/> Unity	<input type="radio"/> Two	<input type="radio"/> Infinity